



Mark Scheme (Results)

Summer 2016

Pearson Edexcel GCSE in Chemistry
(5CH1F) Paper 01
Unit C1: Chemistry in Our World

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General Marking Guidance

- All candidates must receive the same treatment. Examiners must mark the first candidate in exactly the same way as they mark the last.
- Mark schemes should be applied positively. Candidates must be rewarded for what they have shown they can do rather than penalised for omissions.
- Examiners should mark according to the mark scheme not according to their perception of where the grade boundaries may lie.
- There is no ceiling on achievement. All marks on the mark scheme should be used appropriately.
- All the marks on the mark scheme are designed to be awarded. Examiners should always award full marks if deserved, i.e. if the answer matches the mark scheme. Examiners should also be prepared to award zero marks if the candidate's response is not worthy of credit according to the mark scheme.
- For questions worth more than one mark, the answer column shows how partial credit can be allocated. This has been done by the inclusion of part marks eg (1).
- Where some judgement is required, mark schemes will provide the principles by which marks will be awarded and exemplification may be limited.
- When examiners are in doubt regarding the application of the mark scheme to a candidate's response, the team leader must be consulted.
- Crossed out work should be marked UNLESS the candidate has replaced it with an alternative response.

Quality of Written Communication

Questions which involve the writing of continuous prose will expect candidates to:

- Write legibly, with accurate spelling, grammar and punctuation in order to make the meaning clear
- Select and use a form and style of writing appropriate to purpose and to complex subject matter
- Organise information clearly and coherently, using specialist vocabulary when appropriate.

Full marks will be awarded if the candidate has demonstrated the above abilities.

Questions where QWC is likely to be particularly important are indicated (QWC) in the mark scheme, but this does not preclude others.

Question Number	Answer	Acceptable answers	Mark
1 (a) (i)	bar drawn on the graph for nitrogen, to $78 \pm \frac{1}{2}$ small square	Ignore width of line	(1)

Question Number	Answer	Acceptable answers	Mark
1 (a) (ii)	C 21		(1)

Question Number	Answer	Acceptable answers	Mark
1 (b)	An explanation linking (growth of primitive) plants (1) photosynthesis/ (plants take in carbon dioxide and) release/let out/produce oxygen (1)	Allow trees Reject people planting Reject respiration/breathing	(2)

Question Number	Answer	Acceptable answers	Mark
1 (c) (i)	A description linking carbon (atom) (1) and oxygen (atoms) (1)	Ignore numbers of atoms/ symbols molecules = 1 mark max	(2)

Question Number	Answer	Acceptable answers	Mark
1 (c) (ii)	<p>Both answers must come from the same pair.</p> <p>An explanation linking any one of the following pairs</p> <p>burning/combustion (1)</p> <p>(fossil) fuels (1)</p> <p>OR</p> <p>deforestation/cutting down trees (1) reduces { the amount of carbon dioxide taken in/photosynthesis} (1)</p> <p>OR</p> <p>farming /increased {human/animal} population (1)</p> <p>breathing / respiration /exhaling carbon dioxide /decaying/decomposing (1)</p> <p>OR</p> <p>waste/dead organisms (1) decaying/rotting (in landfill) (1)</p> <p>OR</p> <p>limestone (1) decomposed/heated (in lime kiln) (1)</p> <p>OR</p> <p>volcano/volcanic (1) eruption(s) / activity / emits gas (1)</p>	<p>Ignore additional correct answers</p> <p>Maximum 1 mark if additional incorrect answers</p> <p>needs reference to burning Ignore driving/cars/pollution</p> <p>Allow named fossil fuel/any fuel that contains carbon</p> <p>Ignore other human activities</p>	(2)

Question Number	Answer	Acceptable answers	Mark
2(a)	B sodium hydroxide		(1)

Question Number	Answer	Acceptable answers	Mark
2(b)	hydrochloric acid LHS (1) carbon dioxide RHS (1)	Allow correct formulae HCl (1) CO ₂ (1) Reject hydrolic acid, HCL, hCl, carbon oxide, CO2, co2, Co2 and CO ²	(2)

Question Number	Answer	Acceptable answers	Mark
2(c)(i)	glowing splint (1) MP1 relights (1) MP2 MP2 dependent on MP1	Ignore burned out/blown out splint lighted splint burns brighter (2)	(2)

Question Number	Answer	Acceptable answers	Mark
2(c)(ii)	A description linking <ul style="list-style-type: none"> • (volumes of oxygen and/or hydrogen) increase/bigger/goes up (1) • volume of hydrogen (evolved) is always greater than the volume of oxygen (1) <p>OR</p> A correct quantitative description/relationship linking volumes or time and volume <ul style="list-style-type: none"> • volume of hydrogen (evolved) is always double that of oxygen/ volume of oxygen is half that of hydrogen (2) 	just quoting numbers does not show a relationship	(2)

	<ul style="list-style-type: none"> • as time increases the volume (of gas/oxygen /hydrogen) evolved increases OWTTE (2) • as the time doubles the volume(s) (of gas/both gases/oxygen / hydrogen) doubles (2) • volume (of gas evolved) is (directly) proportional to time (2) 		
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Question Number	Answer	Acceptable answers	Mark
2(d)	chlorine (1)	Allow Cl ₂ Reject chloride, Cl, cl, cL	(1)

Question Number	Answer	Acceptable answers	Mark
3(a)	B C ₂ H ₆		(1)

Question Number	Answer	Acceptable answers	Mark
3(b)(i)	fractional distillation (2)	Allow recognisable spellings distillation /fractionation (1)	(2)

Question Number	Answer	Acceptable answers	Mark
3(b)(ii)	B cars aircraft		(1)

Question Number	Answer	Acceptable answers	Mark
3(c)(i)	propane + oxygen (1) {→ / =} carbon dioxide + water (1)	Allow reactants / products in either order, e.g. oxygen + propane LHS C ₃ H ₈ + O ₂ (1) RHS {→ / =} CO ₂ + H ₂ O (1) Ignore incorrect / no balancing Reject incorrect formulae. e.g. h ² O, h ₂ O, H ₂₀ , CO ₂ , CO ² and Co ₂ If a mixture of words and formulae are used, max 1 mark	(2)

Question Number	Answer	Acceptable answers	Mark
3(c)(ii)	C ₃ H ₈ (2)	C and H only with non subscripted/ incorrect / no numbers (1) Allow H ₈ C ₃ /correct structural/ displayed formula Reject 'h' for H	(2)

Question Number	Answer	Acceptable answers	Mark
3(d)	<p>Both marks must come from the same pair.</p> <p>An explanation linking one of the following pairs</p> <p>EITHER carbon monoxide/CO formed (1)</p> <p>toxic/poisonous / restricts the amount of oxygen carried (by the blood)/replaces oxygen in the blood/binds to red blood cells or haemoglobin/causes death (1)</p> <p>OR</p> <p>smoke/soot formed (1)</p> <p>damages lungs/chokes people/breathing difficulties/makes things dirty (1)</p>	<p>Ignore additional correct answers</p> <p>maximum 1 for additional incorrect answers</p> <p>Ignore carbon dioxide</p> <p>Ignore dangerous/harmful Allow the second mark if an incorrect gas e.g. methane Allow second mark if "gas" stated but no name is given Allow kills Allow less energy released</p> <p>Allow carbon</p> <p>Ignore harmful/dangerous Allow blocks fuel jets Allow less energy released</p>	(2)

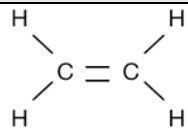
Question Number	Answer	Acceptable answers	Mark
4(a)		if more than one line drawn from/to any metal/method of extraction no marks can be scored for that metal/method of extraction	(3)

Question Number	Answer	Acceptable answers	Mark
4(b)(i)	(zinc oxide +) carbon (→) (1) (zinc +) carbon dioxide / monoxide (1)	Allow correct symbol, C (1) Ignore carbon oxide Allow correct formula, CO ₂ /CO (1)	(2)

Question Number	Answer	Acceptable answers	Mark
4(b)(ii)	reduction		(1)

Question Number	Answer	Acceptable answers	Mark
4(c)	Any two from: <ul style="list-style-type: none"> • (good) electrical conductor (1) • ductile/drawn into a wire (1) • flexible/bendable (1) • unreactive / low reactivity/ resistant to corrosion (1) • high melting point/heat resistant (1) 	Ignore (good) thermal conductor /conductor Ignore malleable Ignore rusting	(2)

Question Number	Answer	Acceptable answers	Mark
4(d)	<p>An explanation linking any two of</p> <ul style="list-style-type: none"> • in pure gold the atoms/particles are all the same size OR in pure gold {layers/sheets/rows} (of atoms) {slide/slip/move} (over each other easily) (1) • (in the alloy there are two types/different sizes of atom/particle) so {structure/layers/sheets/rows} disrupted in alloy (1) • stops/prevents {atoms /particles /layers /sheets /rows} {sliding /slipping /moving} (over one another easily in alloy) (1) 	<p>Allow correct particle diagram(s)</p> <p>Reject molecules <i>once</i></p>	(2)

Question Number	Answer	Acceptable answers	Mark
5(a)1	 <p>(1)</p>	Ignore bond angles	(1)

Question Number	Answer	Acceptable answers	Mark
5(a)2	propene (1)	Reject propane	(1)

Question Number	Answer	Acceptable answers	Mark
5(a)3	C ₄ H ₈ (1)	H ₈ C ₄ numbers must be subscript	(1)

Question Number	Answer	Acceptable answers	Mark
5(b)	<p>A description linking (turns) from orange / red / yellow / brown (1)</p> <p>to colourless/decolourises (1)</p>	<p>Ignore extra incorrect observations</p> <p>Ignore discoloured/clear/transparent</p> <p>Allow stays orange = max 1 mark</p> <p>Reject turns orange</p>	(2)

Question Number	Answer	Acceptable answers	Mark
5(c)	B cracking		(1)

Question Number	Indicative Content	Mark
QWC	<p data-bbox="277 201 367 233">*5(d)</p> <p data-bbox="399 201 1222 264">A description / explanation including some of the following points</p> <p data-bbox="399 302 800 333">properties related to uses</p> <ul data-bbox="440 333 1256 1247" style="list-style-type: none"> <li data-bbox="440 333 1256 596">• poly(ethene) : plastic bags / plastic bottles – flexible/bendable, inert/unreactive, waterproof/weatherproof, light insulation for electrical wires – flexible, bendable, good insulator, waterproof/weather proof, inert/unreactive <li data-bbox="440 621 1256 957">• poly(chloroethene) : window frames / gutters - tough/hard, long-lasting, durable/good insulator, waterproof/weatherproof, inert/unreactive insulation for electrical wires – flexible /bendable, good insulator, waterproof/weather proof, inert/unreactive <li data-bbox="440 982 1256 1247">• poly(tetrafluoroethene): coating for pans - slippery, non-stick, tough, high melting point/heat resistant, inert/unreactive skis - slippery stain proofing fabrics and carpets – slippery containers for corrosive substances – inert/unreactive <p data-bbox="399 1251 760 1283">problems with disposal</p> <p data-bbox="553 1318 1149 1381">Ignore references to cost and/or pollution <u>landfill</u></p> <ul data-bbox="440 1381 1125 1583" style="list-style-type: none"> <li data-bbox="440 1381 1125 1444">• (in landfill) non biodegradable/do not decay/rot/decompose/break down <li data-bbox="440 1444 1125 1476">• waste persists in landfill sites OWTTE <li data-bbox="440 1476 1125 1539">• landfill sites take up space or land / new sites needed when old ones filled <li data-bbox="440 1539 1125 1583">• destroys habitats <p data-bbox="488 1619 602 1650"><u>burning</u></p> <ul data-bbox="440 1650 1097 1814" style="list-style-type: none"> <li data-bbox="440 1650 1097 1713">• (burning polymers) produces carbon dioxide/greenhouse gases <li data-bbox="440 1713 1097 1745">• produces carbon monoxide <li data-bbox="440 1745 1097 1776">• can release toxic/harmful fumes into the air <li data-bbox="440 1776 1097 1814">• produces (toxic) ash <p data-bbox="488 1814 613 1845"><u>recycling</u> (different polymers) need to be separated in order to recycle</p>	(6)

Level	0	No rewardable content
1	1 - 2	<ul style="list-style-type: none"> • a limited description / explanation e.g. one property related to use OR one problem of disposal • the answer communicates ideas using simple language and uses limited scientific terminology • spelling, punctuation and grammar are used with limited accuracy
2	3 - 4	<ul style="list-style-type: none"> • a simple description / explanation e.g. an account including at least one properties related to uses and at least one problems with disposal OR a detailed description of at least two different properties related to uses or at least two problems with disposal • the answer communicates ideas showing some evidence of clarity and organisation and uses scientific terminology appropriately • spelling, punctuation and grammar are used with some accuracy
3	5 - 6	<ul style="list-style-type: none"> • a detailed description / explanation e.g. an account including at least two different properties related to uses AND at least one problem with disposal OR at least one property related to use AND at least two problems with disposal • the answer communicates ideas clearly and coherently uses a range of scientific terminology accurately • spelling, punctuation and grammar are used with few errors

Question Number	Answer	Acceptable answers	Mark
6(a)	(contains) fossils (1)	Allow ammonite	(1)

6(b)	D marble		(1)
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Question Number	Answer	Acceptable answers	Mark
6(c)(i)	thermal decomposition (2)	decomposition (1) Allow recognisable spellings	(2)

Question Number	Answer	Acceptable answers	Mark
6(c)(ii)	calcium oxide + water (1) {→ / =} calcium hydroxide (1)	Allow reactants in either order, e.g. water + calcium oxide LHS CaO + H ₂ O (1) RHS {→ / =} Ca(OH) ₂ (1) Ignore incorrect / no balancing If a mixture of words and formulae are used, max 1 mark	(2)

Question Number	Indicative Content	Mark
QWC	<p data-bbox="256 254 354 289">*6(d)</p> <p data-bbox="375 254 1101 289">A explanation including some of the following points</p> <p data-bbox="418 310 708 346">causes of acid rain</p> <ul data-bbox="418 367 1149 829" style="list-style-type: none">• burning sulfur/coal (in oxygen/air)• sulfur reacts with oxygen• produces sulfur dioxide• acidic gas• in atmosphere dissolves/mixes/absorbs/reacts in rain/clouds/water• to form acid solution/sulfuric acid/sulfurous acid/credit name/correct formula of an acid formed• sulfur dioxide causes acid rain <p data-bbox="418 905 708 940">effects of acid rain</p> <ul data-bbox="418 961 1052 1192" style="list-style-type: none">• damage/erosion/(chemical) weathering of buildings/monuments• acidification of lakes /soil/lowers pH• kills fish/aquatic life• kills/damages trees/ plants/ forests <p data-bbox="418 1268 1008 1325">reduction of the effects of acid rain by limestone</p> <ul data-bbox="418 1346 1122 1654" style="list-style-type: none">• calcium carbonate (from limestone)• acidic gas passed through calcium carbonate in chimney• (calcium carbonate) neutralises/reacts with acidic gases OWTTE• acidic gases prevented from entering the atmosphere	(6)

Level	0	No rewardable content
1	1 - 2	<ul style="list-style-type: none">• a limited explanation e.g. at least one valid idea, e.g. a cause, an effect or a method of reduction of acid rain• the answer communicates ideas using simple language and uses limited scientific terminology• spelling, punctuation and grammar are used with limited accuracy
2	3 - 4	<ul style="list-style-type: none">• a simple explanation e.g. at least two ideas from one aspect OR a mention of a least two aspects• the answer communicates ideas showing some evidence of clarity and organisation and uses scientific terminology appropriately• spelling, punctuation and grammar are used with some accuracy
3	5 - 6	<ul style="list-style-type: none">• a detailed explanation with ideas from all three aspects• the answer communicates ideas clearly and coherently uses a range of scientific terminology accurately• spelling, punctuation and grammar are used with few errors

