UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the October/November 2011 question paper for the guidance of teachers

0620 CHEMISTRY

0620/52

Paper 5 (Practical), maximum raw mark 40

This mark scheme is published as an aid to teachers and candidates, to indicate the requirements of the examination. It shows the basis on which Examiners were instructed to award marks. It does not indicate the details of the discussions that took place at an Examiners' meeting before marking began, which would have considered the acceptability of alternative answers.

Mark schemes must be read in conjunction with the question papers and the report on the examination.

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1

2

(a)	all b	e of results for <i>experiment 1</i> coxes completed correctly (1) dings to 1 d.p. (1) nparable to supervisors (1) ± 3 cm ³	[3]
(b)	all b	e of results for <i>experiment 2</i> poxes completed correctly (1) dings to 1 d.p. (1) apparable to supervisors (1) ± 3 cm ³	[3]
(c)	red	/ brown (1)	[1]
(d)	blue	e / black (1) not : purple to colourless (1) not : clear	[2]
(e)	as a	an indicator / to test for iodine / owtte (1)	[1]
(f)	(i)	experiment 2 (1)	[1]
	(ii)	experiment 2 is 2× volume of experiment 1 (1) allow: almost double	[1]
	(iii)	solution F more concentrated / stronger (1) or converse twice / double (1)	[2]
(g)		value from table result for experiment 1 (1) volume of potassium iodate/iodine (1)	[2]
(h)	(i)	two sources of error e.g. using a measuring cylinder to measure iodate/acid experiment only carried out once going past end point / owtte not: references to human error / temperature / rinsing	[2]
	(ii)	two meaningful improvements related to above e.g. use a pipette / burette repeat experiment	[2]
			[Total: 20]
(a)	(i)	liquid H blue and no smell (1) any reference to smell = 0 pH < 7 (1)	[2]
	(ii)	liquid J colourless (1) not: clear sweet smell / petrol / acetone / owtte (1) not: pungent / strong two layers / immiscible (1) with indicator paper pH 7 or green (1)	[4]

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(b) white precipitate (1) [1] **(c) (i)** blue (1) precipitate (1) on heating black (1) [3] (ii) blue precipitate (1) dissolves / solution (1) deep / royal blue (1) [3] (d) (i) extinguished / goes out / no reaction (1) not: no change / nothing happens [1] (ii) flame / catches fire (1) orange / yellow (1) smokey / sooty (1) max 2 orange / yellow or smokey / sooty flame = (2) reference to glowing splint = 0 [2] **(e)** copper (1) sulfate (1) acidic (1) water present / aqueous (1) max 2 [2]

(f) organic (1) hydrocarbon / non polar (1)

allow: flammable / fuel / alkane not: alcohol

[Total: 20]

[2]