UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS

International General Certificate of Secondary Education

MARK SCHEME for the May/June 2011 question paper for the guidance of teachers

0620 CHEMISTRY

0620/51

Paper 5 (Practical), maximum raw mark 40

Mark schemes must be read in conjunction with the question papers and the report on the examination.

• Cambridge will not enter into discussions or correspondence in connection with these mark schemes.

Cambridge is publishing the mark schemes for the May/June 2011 question papers for most IGCSE, GCE Advanced Level and Advanced Subsidiary Level syllabuses and some Ordinary Level syllabuses.

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(a)	initia to 1	ole of results for Experiment 1 al and final volumes and differences completed correctly (1) decimal place (1) nparable to Supervisor's report (1) ±4 cm ³	[3]	
(b)	b) Table of results for Experiment 2 initial and final volumes and differences completed correctly (1) to 1 decimal place (1) comparable to Supervisor's report (1) ±4 cm ³			
(c)		obles/fizz/effervescence (1) glowing splint (1) relights/brighter (1) ore reference to black	[3]	
(d)	oxy	gen/O ₂ (1)	[1]	
(e)	(i)	colourless not clear to pink/purple (1) or reverse	[1]	
	(ii)	potassium manganate is coloured/is an indicator/owtte (1) allow not an acid/alkali reaction	[1]	
(f)	(i)	experiment 1 (1)	[1]	
	(ii)	experiment 1 two times the volume of experiment 2	[1]	
	(iii)	solution B/potassium manganate more concentrated/stronger (1) or converse two times (1)		
		two times as concentrated = 2	[2]	
(g)		f the value of the result for experiment 2 (1) cm ³ (1) f the volume of C used (1)	[3]	
(h)		rantage easy to use/quick/convenient/ <u>fairly</u> accurate (1) advantage not accurate owtte (1)	[2]	

[Total: 40]

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			IGCSE – May/June 2011	0620	51
2	(a) (i	i) co	lourless (1) smell (1)		[2]
	(ii	i) co	lourless/pale yellow and <u>no</u> smell (1)		[1]
	(b) (i	ye	tches fire/ignites (1) llow/blue (1)		[0]
		ye	llow flame = 2		[2]
	(ii	i) ex	tinguished/no reaction/owtte (1)		[1]
	(c) (i	i) ye	llow/brown/orange/red (1) solution/dissolves (1) not ppt.		[2]
	(ii	i) ye	llow/brown/orange (1)		[1]
	(iii	i) tw	o layers/rings (1) pink/purple top layer (1)		[2]
	(d) (i	i) no	reaction/no change/remains clear/stays same (1)		[1]
	(ii	i) ye	llow (1) precipitate (1)		[2]
	(a) b	brown/yellow/cream/white (1)			
			tate (1)		[2]
	(f) 0	rgani	c (1) alcohol/named alcohol/fuel (1) allow flammable		[2]
	(g) ic	odide/	${ m I^-}$ (1) not iodine		[1]